

# Simulation of Voltage Applied Effect on Carbon-Doped Boron Nitride Nanoribbon Hydrogen Gas Sensing Performance

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**Abstract:** Advanced materials have been widely investigated for numbers of electronics application. The aim is to provide higher device performance and mitigating the short channel effect. In this work, the performance of hydrogen sensing on carbon-doped boron nitride nanoribbon (BC<sub>2</sub>NNR) is investigated in response to different gate voltage applied,  $V_{gs}$ . The simulation is done to see the variation in the density of state (DOS) and the variation in the I – V characteristic. From the simulation, a significant effect is found when the Hydrogen gas was attached to different atom positions within the BC<sub>2</sub>NNR surface. A notable difference in sensitivity was achieved when  $V_{gs}$  is varied from 0 V to 2 V. In fact, a sensitivity of 85.07 % was achieved when the hydrogen molecule was attached to carbon atom. The result presented here may become guideline for experimentalists fabricating BC<sub>2</sub>NNR for gas sensing application.

Keywords: DOS, FET, I-V characteristic, pristine BC<sub>2</sub>NNR

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### 1. INTRODUCTION

Recent advancement in electronics industry have led to a widespread study reported on the use of advance nanomaterial and non - planar devices architecture. Number of efforts have been done particularly on investigating the potential use of nanomaterial such as graphene, carbon nanotube, phosphorene, and boron nitrite (BN) in switching and sensing application. Boron nitrite on one of the materials often studied for various gas sensing application [1]–[3]. A nature of wide band gap in BN has limited use in sensing [4]. However, substitutional and adsorption doping technique has made it possible to engineer BN bandgap allowing possible use in electronics application like gas sensing [5]. A modified version of BN namely hexagonal boron carbon nitride can be formed through alloying process have expand its physical and electrical properties for many devices application.

There is an increased number of studies pertaining to carbon doped boron nitrite (BC<sub>2</sub>N) physical and electrical properties [6], [7]. One of the applications using BC<sub>2</sub>N is gas sensing particularly on hazardous gas [8] which requires a material with high sensitivity, high selectivity, fast response, and rapid recovery time [9], [10]. In a study by Nematollahi et al. [11], they simulated using density functional theory (DFT) to investigate the effect of CO oxidation on Si-doped carbon-doped boron nitride nanosheet (BC<sub>2</sub>NNS) and Si-doped carbon-doped boron nitride nanotube (BC<sub>2</sub>NNT). Their results shows that Sidoped BC<sub>2</sub>NNS offers highly activated substrate to CO oxidation compared to Si-doped BC<sub>2</sub>NNT. In another study, Esrafili et al. [12] performed a simulation study related to the effect of CO and SO<sub>2</sub> molecule on BC<sub>2</sub>N for N<sub>2</sub>O reduction. The edge of BC<sub>2</sub>N is also found to have considerable effect on the electronic properties of the devices [13], [14]. For example, Lu et al. [14] found that H-passivated armchair BC2NNR reveal to have remarkable electronic and magnetic properties. In fact, modification of the material properties by introducing dopant or defect may directly affect these properties. For example, when dopant is introduced on BC<sub>2</sub>N, the charge transfer and binding energy exhibit significant changes [15]. This is supported by a study by Mishra et al. [15] demonstrated that CO<sub>2</sub> binds strongly with an optimal storage of 44 wt % (weight percent) for a monolayer of BCN compared to CH<sub>4</sub>, N<sub>2</sub>, and H<sub>2</sub> adsorption. It is found that when BC<sub>2</sub>N sheets are doped with boron, the adsorption energies for CO<sub>2</sub> and CH<sub>4</sub> considerably increase [15].

Previous works on boron nitrite-based sensing devices result in promising outcome and should be further investigate. This study focusing on evaluating Hydrogen based gas sensor using carbon-doped boron nitride nanoribbon (BC<sub>2</sub>NNR). Number of works has been done related to hydrogen gas sensor [16], [17] where they noted binding energy relation with the presence of electric field. In this work, the hydrogen sensing sensitivity is evaluated by applying different values of back gate voltage. This is to see how the gas molecules reacted upon different gate biased. This is studied through computational simulation and the details of this is presented in the next section.

#### 2. METHODOLOGY

The simulation is performed using Quantum Atomistix Toolkit (ATK) from Synopsys (version T-2022.03-SP1). Armchair BC<sub>2</sub>NNR layer was constructed by using Nanoribbon Plugin Tool in QuantumATK. The number of atoms determining the width of the BC<sub>2</sub>NNR layer is fixed to 12 - dimer line while four unit - cells representing the length, L is obtained by repeating atoms along the C – axis. The hydrogen gas molecules were introduced within the surface by attaching to three different positions namely carbon atom, boron atom and both boron and nitrogen atoms. The temperature was kept constant at 298 K. Figure 1(a - c) presented the structure 12-BC<sub>2</sub>NNR with the presence of H<sub>2</sub> located at carbon, boron, and nitrogen atom respectively. To set - up the device configuration, left and right electrodes were connected at both end of the layers representing the source and the drain region. This is showed in Figure 2. In this study, silicon dioxide, SiO<sub>2</sub> is chosen as gate dielectric, with permittivity of 3.9  $\varepsilon_0$ , a thin metallic layer represent as gate potential was set to 0 V and work function of 5.1  $\varepsilon$ .



Figure 1. Illustration of 12-BC<sub>2</sub>NNR with the presence of H<sub>2</sub> gas in red circle located at (a) carbon atom, (b) boron atom, and (c) both boron and nitrogen atom. White ball, grey ball, pink ball, and blue ball represent hydrogen atom, carbon atom, boron atom, and nitrogen atom, respectively.



Figure 2. Two probes of  $BC_2NNR$  based FET with  $H_2$  gas (a) side view and (b) back view.

In all calculation, the linear combination of atomic orbitals-based density functional theory was used. The Perdew-Zunger formalism of local density approximation was used to compute the exchange-correlation electron. All calculations were performed using the double-zeta polarized basis set. The density mesh cut-off energy was set to 75 Hartree to ensure numerical accuracy. For the electronic structure calculation, the Brillouin zone (BZ) was sampled with a k-point sampling of  $1 \times 1 \times 100$ . The geometries of all atomic structures in the unit cells were initially optimized using the limited-memory Broyden-Fletcher-Goldfarb-Shanno algorithm. geometry optimization is performed using LBFGS to relaxes the structure and finds the minimum energy structure in the modelled layer. To prevent artificial inter-ribbon interactions, a vacuum region of 10 Å was added along the non-periodic boundary.

#### **3. RESULT AND DISCUSSION**

Figure 3 compare the evolution of the projected density of states (DOS) between bare BC<sub>2</sub>NNR and with the presence of hydrogen molecules at three different positions simulated from energy of -2 eV to 2 eV. It can be observed from Figure 3 (a) that bare BC<sub>2</sub>NNR behave typically like one dimensional material where there is a staircase characteristic as the energy increase. This trend can hardly be seen when the hydrogen molecules was attached to BC<sub>2</sub>NNR layers. For example, there is a spike when hydrogen molecules were attached at carbon around energy level of 1.4 eV and 58 eV-1. The DOS on the positive side of the energy level for hydrogen attached to boron and both boron and nitrogen atom have very minimum changes. This can be seen in Figure 3 (c - d)where energy state is found around 2 eV, 1.4 eV, -0.6 eV, -1.5 eV and -1.9 eV.



Figure 3. Device projected density of states (DOS) with (a) bare and presence of  $H_2$  gas (b) carbon atom (c)

boron atom and (d) both boron and nitrogen atoms.

The presence of spike and new energy states upon introduction of gas achieve here has also been showed in other study when they studied propane and butane gas on GNR based FET [18]. This signifies the simulated FET devices demonstrated a considerable change in projected DOS towards the presence of these gas. The focus of this paper is to evaluate the effect of the applied gate biased on the device sensitivity. This is calculated from the changes in the current from the current-voltage (I-V) when hydrogen molecules were presence. The sensitivity is calculated using equation (1) [19].

Sensitivity (%) = 
$$\left|\frac{I - I_0}{I_0}\right| \times 100$$
 (1)

where I is the current with  $H_2$  gas and Io is the current without  $H_2$  gas molecules. The drain to source voltage,  $V_{ds}$  was swept biased, from 0 to 2 V. The sensitivity was calculated and listed in Table 1 below for  $V_{ds} = 2$  V.

Table 1. Comparison of sensitivity for 12-BC<sub>2</sub>NNR based FET at three positions for different  $V_{gs}$  applied taken at  $V_{ds} = 2 \text{ V}.$ 

	Sensitivity (%)		
$V_{gs}(V)$	Carbon	Boron	Boron and nitrogen
0	39.76	-80.50	0.07
0.5	85.07	-80.71	26.45
1	40.46	-90.47	4.30
2	-4.69	-89.74	3.70

For clarity, the  $I_{ds} - V_{ds}$  characteristic was shown from 1.7 V to 2 V as the data below 1.7 is very small. For the three different positions, the characteristic was displayed in Figure 4, Figure 5 and Figure 6 to observe the effect on different gate biased. For all cases, generally there is a nonlinear behavior on the current conduction with the presence of hydrogen molecules as the applied  $V_{gs}$ increase. The average threshold voltage in the presence of H<sub>2</sub> gas showed in back-gate controlled BC<sub>2</sub>NNR are approximately 1.75 V. Appreciable variation in current can be seen particularly at highest V<sub>ds</sub>.



Figure 4. 12-BC<sub>2</sub>NNR based FET in presence of H<sub>2</sub> gas on carbon atom.



Figure 5. 12-BC<sub>2</sub>NNR based FET in presence of  $H_2$  gas on boron atom.



Figure 6. 12-BC<sub>2</sub>NNR based FET in presence of  $H_2$  gas on both boron and nitrogen atoms.

Graphene and BN has shared similar properties and has almost the same with some common characteristic due to the hexagonal structure. Rashid et al. [18] reported, the absorbed target gas molecules behave as charge carrier donors and acceptors for graphene. Hence, the increasing and decreasing of the charge carrier in the current through acted as donors and acceptors [20]. From Figure 4, backgate controlled 12-BC<sub>2</sub>NNR sensor appears to have better sensitivity. H<sub>2</sub> gas exposed at carbon atom may have acted like donors of charge carriers for BC<sub>2</sub>NNR. Current conductivity shows about third quarter increment at  $V_{gs}$  = 0.5 V from 1.33 µA to 2.75 µA with (85.07 %). It then followed by  $V_{gs} = 1$  V and 0 V with 3.42  $\mu$ A (40.46 %) and 1.82 µA (39.76 %), respectively. Whilst a decrease in current conductivity at  $V_{gs} = 2 V$  with 4.69 %, showing  $H_2$ gas exposed at carbon atom have acted like acceptors charge carriers for BC<sub>2</sub>NNR.

On the other hand, H<sub>2</sub> gas exposed at boron atom shown in Figure 5 exhibit a huge drop of current conductivity for all V<sub>gs</sub> with the highest drop of 90.47 % at V<sub>gs</sub> = 1 V, 89.74 % at V<sub>gs</sub> = 2 V, 80.71 % at V<sub>gs</sub> = 0.5 V, and 80.50 % at V<sub>gs</sub> = 0 V. It showed H<sub>2</sub> gas exposed at boron atom may have acted like acceptors and reduced the charge carrier concentration in BC<sub>2</sub>NNR based FET. The most striking different is observed when hydrogen was attached to both boron and nitrogen as shown in Figure 6. For example, at V<sub>ds</sub> = 2 V and V<sub>gs</sub> = 0.5 V, the changes in current are 99.74 %, 83.74 % and 86.01 % compared to V<sub>gs</sub> = 0 V, 1 V and 2 V, respectively. Hydrogen molecules attach to carbon (Figure 4) and both boron and nitrogen (Figure 6) serves better sensing performance compared to boron (Figure 5). Herein, H<sub>2</sub> gas may have acted as donors and increased the charge carrier concentration in BC<sub>2</sub>NNR based FET. This indicates increases with 26.45 % in current conductivity with I<sub>ds</sub> = 1.68  $\mu$ A at V<sub>gs</sub> = 0.5 V.

## 4. CONCLUSION

This paper studied the effect of the applied gate biased on back-gated BC<sub>2</sub>NNR devices. Result indicates significant changes in the projected DOS upon introducing Hydrogen molecules for three different positions namely on carbon, boron, both boron and nitrogen. Generally, there is a current variation achieved when different gate biasing was applied. The device sensitivity calculated showed an upward increase of sensitivity when the hydrogen molecules attached to carbon atom on 12-BC<sub>2</sub>NNR. Significant effect was observed at V<sub>ds</sub> = 2 V where 85.07 % sensitivity is achieve compared to another position. The results presented can further be expanded to evaluate the gate biasing effect for multiple layers of BC<sub>2</sub>NNR.

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#### REFERENCES

- P. Srivastava, N. K. Jaiswal, and V. Sharma, "Firstprinciples investigation of armchair boron nitride nanoribbons for sensing PH3 gas molecules," Superlattices Microstruct., 2014. vol. 73, pp. 350– 358.
- [2] P. Srivastava, V. Sharma, and N. K. Jaiswal, "Adsorption of COCl2 gas molecule on armchair boron nitride nanoribbons for nano sensor applications," Microelectron. Eng., 2015. vol. 146, pp. 62–67.
- [3] P. Srivastava, N. K. Jaiswal, and G. K. Tripathi, "Chlorine sensing properties of zigzag boron nitride nanoribbons," Solid State Commun., 2014. vol. 185, pp. 41–46.
- [4] A. Catellani, M. Posternak, A. Baldereschi, and A. J. Freeman, "Bulk and surface electronic structure of hexagonal boron nitride," *Phys. Rev. B*, 1987. vol. 36, no. 11, pp. 6105–6111.
- [5] M. Sajjad, G. Morell, and P. Feng, "Advance in novel boron nitride nanosheets to nanoelectronic device applications," ACS Appl. Mater. Interfaces, 2013. vol. 5, no. 11, pp. 5051–5056.
- [6] W. H. Lim, A. Hamzah, M. T. Ahmadi, and R. Ismail, "Band gap engineering of BC2N for nanoelectronic applications," Superlattices Microstruct., 2017. vol. 112, pp. 328–338.
- [7] W. H. Lim, A. Hamzah, M. T. Ahmadi, and R. Ismail, "Performance analysis of one dimensional BC2N for nanoelectronics applications," Phys. E Low-Dimensional Syst. Nanostructures, 2018. vol. 102, pp. 33–38.

- [8] Z. Zhang et al., "Hydrogen gas sensor based on metal oxide nanoparticles decorated graphene transistor," Nanoscale, 2015. vol. 7, no. 22, pp. 10078–10084.
- [9] X. Liu, S. Cheng, H. Liu, S. Hu, D. Zhang, and H. Ning, "A survey on gas sensing technology," Sensors (Switzerland), 2012. vol. 12, no. 7, pp. 9635–9665.
- [10] M. Akbari-Saatlu et al., "Silicon nanowires for gas sensing: A review," Nanomaterials, 2020. vol. 10, no. 11, pp. 1–57.
- [11] P. Nematollahi and E. C. Neyts, "A comparative DFT study on CO oxidation reaction over Si-doped BC 2 N nanosheet and nanotube," *Appl. Surf. Sci.*, vol. 439, pp. 934–945, May 2018.
- [12] M. D. Esrafili and N. Saeidi, "Carbon-doped boron nitride nanosheet as a promising catalyst for N 2 O reduction by CO or SO 2 molecule: A comparative DFT study," *Appl. Surf. Sci.*, 2018. vol. 444, pp. 584– 589.
- [13] H. Li, X. Xiao, J. Tie, and J. Lu, "Electronic and magnetic properties tuning of armchair BC2N nanoribbons by edge modification," *Solid State Commun.*, 2017. vol. 257, no. February, pp. 27–31.
- [14] P. Lu, Z. Zhang, and W. Guo, "Magnetism in armchair BC2 N nanoribbons," *Appl. Phys. Lett.*, 2010. vol. 96, no. 13, pp. 7–10.
- [15] A. K. Mishra and S. Mishra, "Tuning of adsorption energies of CO2 and CH4 in borocarbonitrides BxCyNz: A first-principles study," *J. Mol. Graph. Model.*, 2019. vol. 93, p. 107446.
- [16] J. Zhou, Q. Wang, Q. Sun, P. Jena, and X. S. Chen, "Electric field enhanced hydrogen storage on polarizable materials substrates," Proc. Natl. Acad. Sci. U. S. A., 2010. vol. 107, no. 7, pp. 2801–2806.
- [17] H. Guo et al., "CO2 capture on h -BN sheet with high selectivity controlled by external electric field," J. Phys. Chem. C, 2015. vol. 119, no. 12, pp. 6912– 6917.
- [18] M. H. Rashid, A. Koel, and T. Rang, "Simulations of graphene nanoribbon field effect transistor for the detection of propane and butane gases: A first principles study," Nanomaterials, 2020. vol. 10, no. 1.
- [19] Salih, E., & Ayesh, A. I. (2021). Pt-doped armchair graphene nanoribbon as a promising gas sensor for CO and CO2: DFT study. Physica E: Lowdimensional Systems and Nanostructures, 125, 114418.
- [20] F. Yavari and N. Koratkar, "Graphene-Based Chemical Sensors," 2012.